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## LETTER TO THE EDITOR

## Proton glass behaviour in a solid solution of $\gamma$ -irradiated deuterated betaine phosphate<sub>0.15</sub> betaine phosphite<sub>0.85</sub>

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**Abstract.** Measurements of the dielectric permittivity are reported for a deuterated solid solution of  $\gamma$ -irradiated antiferroelectric (betaine phosphate)<sub>0.15</sub> ferroelectric (betaine phosphite)<sub>0.85</sub> at frequencies 20 Hz <  $\nu$  < 1 MHz. The freezing phenomena in DPB<sub>0.15</sub>DBPI<sub>0.85</sub> revealed the characteristics of a transition into a dipolar glass state. The activation energy was found to be  $E_b = 311.6$  K (0.027 eV). The Kutnjak model showed a non-typical glass behaviour with an estimated glass temperature of 55.4 K.

Betaine phosphate (BP; (CH<sub>3</sub>)NCH<sub>2</sub>COOH<sub>3</sub>PO<sub>4</sub>) and betaine phosphite (BPI; (CH<sub>3</sub>)NCH<sub>2</sub> COOH<sub>3</sub>PO<sub>3</sub>) are molecular crystals of the amino acid betaine and phosphoric and phosphorous acids, respectively. In both compounds the inorganic components ( $PO_4$  or PO<sub>3</sub> groups) are linked by hydrogen bonds to form quasi-one-dimensional chains [1]. BP exhibits a ferroelastic phase transition at about 365 K followed by two phase transitions at 86 and 81 K [1]. Antiferroelectric order is established at  $T_c = 86$  K [2]. At this temperature the O-H...O bonds order along the one-dimensional chains and the chains are linked antiferroelectrically [3]. At 355 K, BPI transforms into an elastically ordered state and exhibits ferroelectric order below  $T_c = 216$  K [1,4]. The two almost isostructural compounds form solid solutions at any concentration [4]. It has been shown recently that at intermediate concentrations the long-range electric order is suppressed [4] and no spontaneous polarization occurs [5,6]. The mixed crystal BP<sub>0.4</sub>BPI<sub>0.6</sub> exhibits relaxational behaviour typical for an orientational glass state [7], with a hindering barrier  $E_b = 252$  K (0.022 eV). For BP<sub>0.15</sub>BPI<sub>0.85</sub> no anomaly in  $\varepsilon'$  indicating the polar phase transition can be detected down to the lowest temperatures [5,6]. The most probable relaxation rate follows an Arrhenius law  $\tau = \tau_0 \exp(E_b/kT)$ , with  $\tau_0 = 2.4 \times 10^{-13}$  s and  $E_b = 518$  K (0.045 eV) [8]. The activation energy  $E_b$  is significantly higher than in BP<sub>0.4</sub>BPI<sub>0.6</sub> [7]. An external bias field E = 0.57 kV cm<sup>-1</sup> significantly lowers the dielectric permittivity and changes the temperature behaviour of  $\varepsilon''$  of BP<sub>0.15</sub>BPI<sub>0.85</sub> [8]. The relaxation time with bias field follows an Arrhenius law with parameters  $\tau_0 = 7.1 \times 10^{-13}$  s and  $E_A = 488$  K (0.042 eV). 1 Mrad  $\gamma$ -irradiation increases the activation energy to 513 K (0.044 eV) [9].

Deuteration of hydrogen bonded ferroelectrics leads to significant changes of the dielectric properties and shifts the phase transition temperature to higher values [10–16]. This isotope effect has already been studied in deuterated crystals of the betaine family, namely betaine phosphate (DBP) and betaine phosphite (DBPI) [12–16]. The low-frequency dielectric measurements of DBPI showed [15, 16] that the ferroelectric phase transition is

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**Figure 1.** The temperature dependences of (a)  $\varepsilon'$  and (b)  $\varepsilon''$  of DBP<sub>0.15</sub>DBPI<sub>0.85</sub> at the following frequencies (kHz):  $\Box$ , 0.2;  $\bigcirc$ , 2;  $\triangle$ , 20;  $\bigtriangledown$ , 200;  $\diamondsuit$ , 1000.

shifted up to 297 K. EPR studies revealed that the ferroelectric phase transition is related to the simultaneous ordering of deuterons in the O-D...O bonds [15].

In this letter we present the results of dielectric investigations of  $\gamma$ -irradiated DBP<sub>0.15</sub>DBPI<sub>0.85</sub> samples.

DBP<sub>0.15</sub>DBPI<sub>0.85</sub> crystals were grown by controlled evaporation from D<sub>2</sub>O solution containing betaine and H<sub>3</sub>PO<sub>3</sub> and H<sub>3</sub>PO<sub>4</sub>. By analogy with DBPI [15] one expects that only the protons in the O–H...O bonds of the inorganic H<sub>3</sub>PO<sub>3</sub> and H<sub>3</sub>PO<sub>4</sub> groups should be replaced by deuterons. Crystals for measurements were irradiated by  $\gamma$ -rays with a dose of 1 Mrad. For the dielectric spectroscopy gold-plated single crystals were oriented along the monoclinic *b* axis. The complex dielectric constant  $\varepsilon^* = \varepsilon' - i\varepsilon''$  was measured by a capacitance bridge HP4284A in the frequency range 20 Hz–1 MHz. For the temperature dependent measurements a Leybold VSK-4-320 cryostat was used. All measurements were performed on heating at a rate of about 0.1 K min<sup>-1</sup> in the phase transition region. The sample was a thin plate of 0.25 mm thickness and 8.5 mm<sup>2</sup> area coated with gold electrodes.

A Curie–Weiss law is valid for  $\varepsilon'$  in the temperature range 300–200 K. At lower temperatures the deviation from the Curie–Weiss law is very significant. A similar behaviour



**Figure 2.** The frequency dependences of (a)  $\varepsilon'$  and (b)  $\varepsilon''$  of DBP<sub>0.15</sub>DBPI<sub>0.85</sub> at the following temperatures (K):  $\triangle$ , 40;  $\bigcirc$ , 50;  $\Box$ , 60;  $\bigtriangledown$ , 70.

observed in  $Rb_{1-x}(NH_4)_x H_2AsO_4$  has been explained by random field freezing [17, 18]. This means that in  $DBP_{0.15}DBPI_{0.85}$  random field freezing occurs at temperatures lower than 200 K. At temperatures lower than 80 K dispersion effects dominate the dielectric response (figure 1). The maximum value of  $\varepsilon'$  is 28 and is much lower than in the non-deuterated sample [8, 9].

The frequency dependences of  $\varepsilon'$  and  $\varepsilon''$  at fixed temperatures provide clear evidence that the  $\varepsilon''$  frequency dependence is much broader than 1.14 decades as it should be for the Debye dispersion (figure 2). The Cole–Cole diagram (figure 3) also shows that there is a wide distribution of relaxation times. The freezing phenomena in DBP<sub>0.15</sub>DBPI<sub>0.85</sub> reveal the characteristics of a transition into a dipolar glass state: the slowing down of the dipolar degrees of freedom exhibits a broad distribution of the relaxation rates, with the width of the distribution exceeding by orders of magnitude the width of a monodispersive Debye process [19, 20]. Dipolar glasses have been studied in detail: the most prominent examples are K<sub>1-x</sub>Li<sub>x</sub>TaO<sub>3</sub> [21], Rb<sub>1-x</sub>(NH<sub>4</sub>)<sub>x</sub>H<sub>2</sub>PO<sub>4</sub> [22], and Rb<sub>1-x</sub>(NH<sub>4</sub>)<sub>x</sub>H<sub>2</sub>AsO<sub>4</sub> [17, 18]. In orientational glasses (OGs) the reorienting moments freeze-in in random configurations. The interplay of site disorder and frustrated interactions is responsible for the freezing



**Figure 3.** Cole–Cole diagrams of DBP<sub>0.15</sub>DBPI<sub>0.85</sub> at the following temperatures (K):  $\triangle$ , 40;  $\bigcirc$ , 50;  $\Box$ , 60;  $\bigtriangledown$ , 70.

transition, which bears similarities to the spin-glass transitions in dilute magnetic systems [23] and to relaxational dynamics in canon-ical glasses [24]. The experimental data were fitted with the Cole–Cole function [25]:

$$\varepsilon^* = \varepsilon_{\infty} + \Delta \varepsilon / \left[ 1 + (i\omega\tau)^{1-\alpha} \right]$$

where  $\Delta \varepsilon$  is the relaxator strength,  $\tau$  is the most probable relaxation time,  $\alpha$  is the distribution coefficient,  $\varepsilon_{\infty}$  is the contribution of all higher-frequency modes to the dielectric permittivity, and  $\omega = 2\pi v$  is the angular velocity. The data calculated using the Cole–Cole formula are shown as solid lines in figures 2 and 3. The temperature dependences of the fit parameters  $\Delta \varepsilon$ ,  $\alpha$ , and  $\tau$  are shown in figure 4.

When  $\alpha > 0.5$  the relaxation rates are distributed over three decades. Such a wide distribution of relaxation rates can mean that non-equilibrium effects or quantum fluctuations play a significant role at these temperatures where  $\alpha$  reaches high values. Deviations from purely thermally activated processes at low temperatures might indicate the increasing importance of tunnelling transitions at low temperatures as in KI:NH<sub>4</sub>I [26]. The activation energy obtained,  $E_b = 311$  K (0.027 eV), is similar to that in pure BPI as shown by <sup>1</sup>H ENDOR experiment [27] and dielectric measurements [28] and in BP<sub>0.15</sub>BPI<sub>0.85</sub> [8,9]. In the glassy state the protons are frozen-in at random in the double-minimum potentials, as in RADP [22], along the one-dimensional chains without any long-range order. The distributions of relaxation times are symmetrically shaped and can be explained in terms of a distribution of energy barriers [25, 26]. The cusp of the dispersion strength  $\Delta\varepsilon$  indicates the temperature  $T_a \approx 50$  K at which freezing of random bonds occurs (figure 4).

The observed experimental data were used for calculations with the model recently suggested by Kutnjak *et al* [29]. For this purpose the variable  $\delta(v) = [\varepsilon'(v) - \varepsilon_{\infty}]/[\varepsilon_s - \varepsilon_{\infty}]$  is introduced with  $\varepsilon_s$  and  $\varepsilon_{\infty}$  denoting static and high-frequency permittivity. By determining the value of the frequency  $v_{\delta}$  for a given  $\delta$  and for all accessible temperatures it is possible to construct an Arrhenius plot, the so called  $\delta$ -plot shown in figure 5. Notable cases are  $\delta \rightarrow 0$  and  $\delta \rightarrow 1$  which correspond to the high- and low-frequency ends of the spectrum



**Figure 4.** The temperature dependences of the fit parameters  $\alpha$ ,  $\tau$  and  $\Delta \varepsilon$ .



**Figure 5.** An Arrhenius representation ( $\delta$ -plot) of data obtained from the reduced dieletric constant  $\delta(\nu) = [\varepsilon'(\nu) - \varepsilon_{\infty}]/[\varepsilon_s - \varepsilon_{\infty}]$  at the following  $\delta$ -values: (a)  $\Box$ , 0.9;  $\bigcirc$ , 0.8;  $\triangle$ , 0.7;  $\nabla$ , 0.6;  $\Diamond$ , 0.5 and (b) the  $T_0$  dependence on  $\delta$ .

of relaxation times. Most interesting is of course the limit  $\delta \rightarrow 1$  which reveals the temperature evolution of the slowest element in the spectrum. From figure 5 it is evident that for  $\delta < 0.95$  the curves develop considerable curvature. The solid lines in this figure have been calculated using the Vogel–Fulcher expression  $v_{\delta} = v_0 \exp(-B/(T - T_0))$  with

a  $\delta$ -dependent Vogel temperature [29]. In figure 5(b) we show  $T_0$  as a function of  $\delta$  for 0.5 <  $\delta$  < 0.9. It is clearly seen that for  $\delta \rightarrow 1$  the Vogel temperature extrapolates to 55.4 K. It has to be noted that Vogel temperatures obtained from figure 5(a) are negative whereas in BP<sub>0.4</sub>BPI<sub>0.6</sub> it had a positive value [30]. This fact can be explained as follows: the case of BP<sub>0.4</sub>BPI<sub>0.6</sub> was an ergodic glass state, and in our case we observe a non-ergodic glass state.

Finally, one can conclude that DBP<sub>0.15</sub>DBPI<sub>0.85</sub> exhibits an orientational glass state at low temperatures where the protons are frozen-in along the one-dimensional chains. The loss peaks are broad, indicating a wide distribution of relaxation rates. The value of activation energy clearly manifests that the orientational glass state is related to the proton order.

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